### Nipomo Community Services District CONSUMER CONFIDENCE REPORT – 2009 WATER QUALITY DATA

The table below lists all the drinking water contaminants that we detected during the 2009 calendar year. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. Unless otherwise noted, the data presented in this table is from testing done January 1 - December 31, 2009. The State requires us to monitor for certain contaminants less than once per year because the concentrations of those contaminants are not expected to vary significantly from year to year. Some of the data, though representative of the water quality, is more than one year old.

#### Terms & abbreviations used below:

- Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.
- Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency.
- Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

- Maximum Residual Disinfectant Level (MRDL): The level of a disinfectant added for water treatment that may not be exceeded at the consumer's tap.
- Maximum Residual Disinfectant Level Goal (MRDLG): The level of a disinfectant added for water treatment below which there is no known or expected risk to health. MRDLs are set by the U.S. Environmental Protection Agency.
- Regulatory Action Level (AL): The concentration of a contaminant, which, if exceeded, triggers treatment or other requirements that a water system must follow.
- Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements
- Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, order, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.
- NA not applicable ND: not detectable at testing limit NS: no standard or not regulated MFL: million fibers per liter
- NTU: Nephelometric Turbidity Units pCi/l: picocuries per liter (a measure of radiation) ppb: parts per billion or micrograms per liter (µg/L) ppm: parts per million or milligrams per liter (mg/L) ppq: parts per quadrillion or picograms per liter (pg/L) ppt: parts per trillion or nanograms per liter (ng/L)

Detected Contaminants	Units	No. of Samples Collected	No. Sites Exceeding AL	90 <sup>th</sup> Percentile Level	∍ AL	PHG	Typical Sources of Contaminants	
Lead (Pb)	ppb	27 (2009)	0	0.50	15	0.2	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers, erosion of natural deposits	
Copper	ppm	27 (2009)	1	0.193	1.3	0.17	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives	
Primary Drinkin	g Water S	Standards	(PDWS)				·	
Detected Contaminants	Units	MCL	PHG (MCLG)	Averaç	Results ge Range	Тур	Typical Sources of Contaminants	
Aluminum (Al)	ppm	1	0.6	0.003	ND - 0.03 (2008)	trea	Erosion of natural deposits; residue from some surface water treatment processes	
Arsenic (As)	ppb	10	NA	9.0	8 - 10 (2009)		Erosion of natural deposits; runoff from orchards, glass and electronics production wastes	
Barium (Ba)	ppm	া	2	0.03	0.03 - 0.05 (2008)	Dis ero	Discharge from oil drilling wastes and from metal refineries; erosion of natural deposits	
Cadmium (Cd)	ppb	5.0	0.04	0.5	ND - 1 (2008)	der che	Internal corrosion of galvanized pipes; erosion of natural deposits; discharge from electroplating and industrial chemical factories, and from metal refineries; runoff from waste batteries and paints	
Chromium	ppb	50.0	NA	0.7	(2008) er		scharge from steel and pulp mills and chrome plating;: osion of natural deposits	
Fluoride (F)	ppm	2	1	0.25	ND - 0.5 (2008)	stro	osion of natural deposits; water additive that promotes ong teeth; discharge from fertilizer and aluminum factories	
Nitrate (NO3)	ppm	45	45	7.4	ND - 17 (2009)		noff and leaching from fertilizer use; leaching from septic lks and sewage; erosion of natural deposits	
Selenium (Se)	ppb	50	NA	6.0	ND - 10 (2008)	erc che	scharge from petroleum, glass, and metal refineries; osion of natural deposits; discharge from mines and emical manufacturers; runoff from livestock lots (feed ditive)	
Gross Alpha	pCi/L	15	NA	1.3	1 - 1 (2009)	Erc	osion of natural deposits	
Uranium	pCi/L	20	0.5	3.0	0.2 - 6 (2008)	Erc	osion of natural deposits	
Chloride	ppm	500	NA	59	41 - 97 (2009)	Ru	noff/leaching from natural deposits; seawater influence	

Detected Contaminants	Units	MCL	PHG (MCLG)	Results Average Range		Typical Sources of Contaminants
Color (Unfiltered)	Units	15	NA	5	ND - 20 (2008)	Naturally-occurring organic materials
Corrosivity (Langlier Index)	NA	> 0	NA	-0,5	-2 - 0.4 (2008)	Natural or industrial-influenced balance of hydrogen, carbon and oxygen in the water; affected by temperature, other factors
Iron (Fe)	ppb	300	NA	110	ND - 600 (2009)	Leaching from natural deposits; industrial wastes
Manganese (Mn)	ppb	50	NA	12	ND - 100 (2009)	Leaching from natural deposits
Odor Threshold at 60°	TON	3	n/a	0.2	ND – 4 (2008)	Naturally-occurring organic materials
Specific Conductance	umhos/ cm	1600	NA	845	358 - 1340 (2008)	Substances that form ions when in water; seawater influence
Sulfate (SO4)	ppm	500	NA	189	80.0 - 322 (2009)	Runoff/leaching from natural deposits; industrial wastes
TDS	ppm	1000	NA	574	410 - 760 (2009)	Runoff/leaching from natural deposits
Zinc (Zn)	ppm	5	NA	0.004	ND - 0.06 (2008)	Runoff/leaching from natural deposits
			U	nregula	ted Contan	ninants
Detected Contaminants	Units	Action Level	Results Average Range			Typical Sources of Contaminants
Boron	ppm	1000	0.06		ND - 0.1 (2009)	Some men who drink water containing boron in excess of action level over many years may experience reproductive effects, based on studies in dogs
Vanadium	ppm	50	0.0	08	ND - 0.01 (2008)	The babies of some pregnant women who drink water containin vanadium in excess of action level may have an increased risk of developmental effects, based on studies in laboratory animal
Bromodichloromethane	ppb	n/a	0,08		ND - 0.7 (2008)	n/a
Sampling Results for	or Sodi	um and	Hardn	ess		
Detected Contaminants	Units	Action Level	Avera	Resu	lts Range	Typical Sources of Contaminants
Sodium	ppm	NS	6	1	48 - 93 (2009)	Sodium refers to the salt present in the water and is generally naturally occurring
Total Hardness (as CaCO3)	ppm	NS	29	94	57 - 528 (2008)	Hardness is the sum of polyvalent cations present in the water, generally magnesium and calcium. The cations are usually naturally-occurring

Items shaded are greater than MCL or AL.

For Arsenic (As) results above 5 ppb up to and including 10 ppb: While your drinking water meets the federal and state standard for arsenic, it does contain low levels of arsenic. The arsenic standard balances the current understanding of arsenic's possible health effects against the costs of removing arsenic from the drinking water. The U.S. Environmental Protection Agency continues to research the health effects of low levels of arsenic, which is a mineral known to cause cancer in humans at high concentration and is linked to other health effects such a skin damage and circulatory problems.

<u>About our Color (Unfiltered)</u>: Color was found at levels that exceed the secondary MCL. The color MCL was set to protect you from unpleasant aesthetic affects due to color. Violating this MCL does not pose any risk to public health.

About our Corrosivity (Langlier Index): Corrosivity less than 0 indicates you water may be corrosive to the plumbing and fixtures. The Corrosivity MCL was set to protect you against unpleasant aesthetic affects such as color, taste and odor. Violating this MCL does not pose a risk to public health.

About our Iron (Fe): Iron was found at levels that exceed the secondary MCL. The Iron MCL was set to protect you against unpleasant aesthetic affects such as color, taste, odor, and the staining of plumbing fixtures (e.g. tubs and sinks) and clothing while washing. Violating this MCL does not pose any risk to public health.

<u>About our Manganese (Mn)</u>: Manganese was found at levels that exceed the secondary MCL. The Manganese MCL was set to protect you against unpleasant aesthetic affects such as color, taste, odor, and the staining of plumbing fixtures (e.g. tubs and sinks) and clothing while washing. Violating this MCL does not pose any risk to public health.

About our Odor Threshold at 60°C): Odor was found at levels that exceed the secondary MCL. The Odor ML was set to protect you against unpleasant aesthetic effects such as color, taste, odor, and the staining of plumbing fixtures (e.g. tubs and sinks) and clothing while washing. Violating this MCL does not pose any risk to public health.

**Compliance with Other Regulations:** The State requires us to test our water on a regular basis to ensure its safety. In 2009, we met all sampling, treatment and reporting requirements.

Check the NCSD website regularly for water conservation tips, news about free gardening and irrigation workshops, information on Household Hazardous Waste disposal, recycling and more!

# THANK YOU FOR SAVING WATER!

CONSUMER CONFIDENCE **QUALITY DATA** REPORT 2009 WATER

# Nipomo Community Services District



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# NIPOMO COMMUNITY SERVICES DISTRICT CONSUMER CONFIDENCE REPORT Water Quality Data 2009

This brochure is a snapshot of the quality of

the water that we provided last year

importante sopre su ugua. Tradúzcalo ó hable con alguien que entienda bien. informe contiene información beber. muy

Included are details about where your water comes from, what it contains, and how it compares to State standards. We are committed to providing you with this information to keep you informed about your water supply.

For more information about your water, you may call (805) 929-1133 and ask to speak with a member of the District's professional staff. The

Vice President President Director Director Director Larry Vierheilig Jim Harrison Mike Winn Bill Nelson Ed Eby

## NCSD Elected Board of Directors

especially 7. Buy high-efficiency appliances, clothes-washer and dishwashe

automatic shut-off nozzle when washing your car, or take your car to a carwash that with a bucket and a hose take your recycles water. 6. Use

to clean hose) 5. Use a broom (not a driveways, decks and patios. a broom (not

and AM, Irrigate between 10 PM and 6 avoid irrigating when it is windy.

irrigation he timer system a habit, including changing settings as seasons change. Make monthly tune-ups of your 8 3

2. Go low-flow! Replace showerheads, faucets and toilets with water-efficient, lowflow fixtures.

Find and repair leaks.



a number of ways to save water, There are a number of war but they all begin with you!

The most important component of wa efficiency is you! You are the one who rea a water-efficiency tip and then puts it a water-efficiency tip and then puts it action. Without your action, tips on sav water remain just words written on a page.

a.gov/Library/water conservation/ visit http://ncsd.oa 25-tips-Web.

on using water more efficiently, For 25 tips

continue to become more scarce, we will need to continue saving water every way we can. If we do not conserve, we are pouring can reduce your water consumption by taking just a few simple steps. down the drain. sonrces water new and money since However

d news is NCSD's customers have ad their water consumption by 14% 07, and by 17% since 2003. This is an amount of conservation, and our ris deserve to be proud of this accomplishment decreased the since 2007, a significant a customers The good

Water conservation is a key link connecting the supply and cost of water. It also is the most cost-effective way to decrease our demand for new sources of water. As new sources of water become more scarce, the sources of water become more scarce, the price for water goes up, and that increase in price is passed on to the customer.

\$ave Water, \$ave Money!

District holds public meetings the second and fourth Wednesday of each month at 9:00 a.m. at 148 South Wilson Street in Nipomo, California. Meeting agendas and breaking District news can be accessed on the District's website at www.ncsd.ca.gov

sources: the Nipomo Mesa Sub-Basin of the San Maria Valley Groundwater basin, and the Nipomo Valley Basin via nine wells. Your water comes from two distinct groundwater sources: the Nipomo Mesa Sub-Basin of the Santa

sources was completed June 2001. Our water resources are considered most vulnerable to the following activities: pesticides and fertilizers associated with golf courses and agriculture, low and high-density septic systems, sewer collection systems, and wastewater treatment plants. A copy of the complete assessment may be viewed at the District office. An assessment of the District's drinking water

dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity. Sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, spring, and wells. As water travels over the surface of the land or through the ground, it

water include: Contaminants that may be present in any source

- bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations and wildlife. Microbial contaminants, such as viruses and
- metals, which can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining or farming Inorganic contaminants, such as salts and
- Pesticides and herbicides, which may come from a variety of sources: agriculture, urban

stormwater runoff, and residential uses.

- production and mining activities Radioactive contaminants, which can be occurring or the result of oil
- Organic chemical contaminants, including synthetic and volatile organic chemicals, which are byproducts of industrial processes and petroleum production, and can come from gas stations, urban stormwater runoff, and septic systems

Services (Department) prescribe regulations which limit the amount of certain contaminants in water provided by public water systems. Department regulations also establish limits for contaminants in bottled water that must provide the same protection for public health. In order to ensure that tap water is safe to drink, USEPA and the California Department of Health

contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the EPA's Safe Drinking Water Hotline 1-800-426-4791. **Drinking water, including bottled water**, may reasonably be expected to contain at least small amounts of some contaminants. The presence of

population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy persons who have undergone organ transplants, people with HIV/AIDS or other immune system infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline: 1-800-426-4791. disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care provider. EPA/CDC guidelines on appropriate means to lessen the risk of Some people may be more vulnerable to contaminants in drinking water than the general

